

helix has shown that there is a sequence dependent cationic localization toward the purine-rich run within the TAR duplex. A region of high ion affinity agrees very well with the position of the X-ray determined divalent cations within a fragment from the HIV-1 TAR RNA. We show that a unique sequestration of ions within the core helix occurred independently of a nucleotide bulge and solely based on sequence of the helix. Our results suggest a high propensity toward purine dependent colocalization of one to two cations distinct from those performing phosphate backbone screening.

## 225-Pos

### Computational Exploration of Thermodynamics and Kinetics of Mobile Ions Around RNA Duplex

Serdal Kirmizialtin, Ron Elber.

The University of Texas Austin, Austin, TX, USA.

Atomically detailed distributions of ions around an A-form RNA are computed. Different mixtures of monovalent and divalent ions are considered explicitly. Studies of tightly bound and of diffusive (but bound) ions around 25 base pairs RNA are conducted in an explicit solvent. Replica exchange simulations provide detailed equilibrium distributions with moderate computing resources (20 nanoseconds of simulation using 64 replicas). Magnesium ion distributions show significant near-RNA binding while sodium ion distributions are more diffusive. Predicted binding sites of at the RNA surface are in accord with structures from crystallography. Electric field relaxation is investigated. The relaxation due to solution rearrangements relaxes in tens of picoseconds, while the contribution of RNA tumbling continues to a few nanoseconds. Negative mobile ions can be found near the RNA but must be assisted by proximate and mobile cations. At distances larger than 16 Å from the RNA center, a continuum model of RNA charge density and solution becomes accurate. At shorter distances, the structure of RNA (and ions) has significant impact on the pair correlation functions.

## 226-Pos

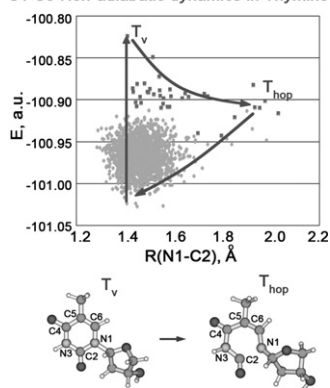
### Photochemistry of DNA Fragments Via Semiclassical Nonadiabatic Dynamics

Anastassia Alexandrova<sup>1</sup>, John Tully<sup>1</sup>, Giovanni Granucci<sup>2</sup>.

<sup>1</sup>Yale University, New Haven, CT, USA, <sup>2</sup>Università di Pisa, Pisa, Italy.

Forming upon absorption of a UV photon, excited states of DNA are subject to nonadiabatic evolution. Though photo-excited DNA mostly undergoes internal conversion back to the ground state, various routes of mutagenesis are also possible. Ultimately, the accumulation of errors in the genome can result in cancer. Here, nonadiabatic processes following the formation of the first singlet excited states, S1, in ten different small DNA fragments have been investigated: four single 4'-H-nucleosides, two Watson-Crick base pairs, and four nucleotide quartets. Simulations were done via the nonadiabatic direct trajectory surface hopping semiclassical dynamics. The electronic wavefunction was obtained with configuration interaction, based on the semiempirical PM3 Hamiltonian with fractional orbital occupation numbers. The evolution of the electronic wavefunction was governed by the time-dependent Schrödinger's equation with a locally-diabatic representation, intrinsically stable near surface crossings. The nuclei evolved on adiabatic potential energy surfaces, as prescribed by classical Newtonian dynamics. The "fewest switches" surface hopping algorithm coupled the quantum and classical parts of the system. The dynamics simulations revealed several routes of nonadiabatic relaxation in these systems, which were not reported previously.

S1-S0 Non-adiabatic dynamics in Thymine



## 227-Pos

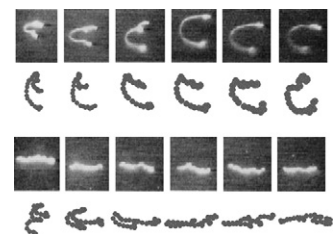
### Application of Reptation Model on Brownian Dynamics for Electrophoresis of Single DNA in Polymer Solution

Seungtae Kang, Byung Jun Yoon.

Postech, Pohang, Korea, Republic of.

Brownian dynamics(BD) simulation is performed to study electrophoretic motion of a single DNA molecule in polymer solution. When a DNA is forced to pass through pores in polymer solution under electrophoresis, the motion of DNA is strongly influence by surrounding entangled polymer molecules. We following the concept in the reptation model to represent the dynamics of DNA in polymer solution. Using the cubic Bezier spline, we manifest the con-

tour of DNA to apply the constraint force from entangled polymer molecules surrounding the DNA. U-shaped, I-shaped migration, and periodic motions of DNA corresponding to each concentration of polymers solution under DC field, and the dynamics of DNA under AC field are simulated. We derive electrophoretic mobility using BD model with the constraint force to compare with experiment. We make the empirical correlation of the constraint force with concentration of polymer solution.



## 228-Pos

### Partitioning of the Elastic Energy in Protein-Dna Chimeras

Andrew Wang, Chiao-Yu Tseng, Biljana Rolih, Alex J. Levine, Giovanni Zocchi.

UCLA, Los Angeles, CA, USA.

We synthesize Protein-DNA chimeras where a DNA molecular spring mechanically perturbs the conformation of the protein. We measured the elastic energy stored in one such molecule, consisting of the enzyme Guanylate Kinase coupled to a 60 bp DNA spring. From these measurements, the response of the protein in terms of its enzymatic activity, and a mechanical model of the DNA spring we deduce that, in this case, most of the elastic energy of the molecule is stored in the DNA spring. Thus the DNA spring is "softer" than the protein.

## 229-Pos

### Self-Assembly in a Model Amphiphile System

Lorna Dougan.

University of Leeds, Leeds, United Kingdom.

The physical origin of the large and negative excess entropy of mixing of alcohols and water remains controversial. In contrast to standard explanations that evoke concepts of water structuring, recent work has shown that, at ambient conditions, it can be quantitatively explained in terms of molecular scale partial demixing of the two components. Here, we estimate the negative excess entropy of aqueous methanol at low temperature and high pressure using experimentally-derived structural data and a recently introduced cluster model. On cooling to 190 K the cluster sizes increase, but the change in negative excess entropy, which according to this method of calculation depends on the surface area to volume ratio of the clusters, is not significant, suggesting that the topology of the clusters must change with decreased temperature. On compression the cluster sizes also increase, and the negative excess entropy is now positive, suggesting an even more pronounced change in cluster topology with increased pressure. This work suggests that it is the amphiphilic nature of a molecule that determines aggregation and self-assembly processes in aqueous solution.

The results therefore give useful insight into the processes of cold and pressure denaturation of proteins.

## 230-Pos

### Hydrophobic and Hydrophilic Interactions

David V. Svintradze.

OCMB Philips Institute and Institute for Structural Biology&Drug Discovery, Virginia Commonwealth University, Richmond, VA, USA.

We are trying to understand mysteries of nature by looking at extra large (Cosmology and Astrophysics) and extra small distances (Weak/Strong interactions High energy Physics) while it can be seen in the physics of life systems. In previous presentations I developed some theoretical vision about conformational motion and non-equilibrium dancing of biological macromolecules by giving definitions of non-equilibrium entropy and geometrical motion. Those two key definitions and formulations are believed to be enough to answer the questions what is big energetic fluctuation is induced by in non-equilibrium systems and how the life system functions properly under that fluctuations. Self-organization is dynamic process where hydrophobic-hydrophilic interactions take a crucial role. Non-equilibrium dancing may induce 'hydrophobic-hydrophilic' waves which may be felt by other molecules. One may think that the suggestion about conformational motion may complicate quantitative and qualitative description of hydrophobic-hydrophilic interactions. Nevertheless, geometrical motion itself indicates changes of hydrophobicity of the surfaces and can be completely described if it is taken into account dynamic processes of the surfaces solvent interactions. Since, in solvents, we have large number of interacting molecules statistical physics supposed to have crucial role in describing those dynamic processes but task is complicated by non-equilibrium nature of the processes. Fluctuation theorem guarantees reversibility of non-equilibrium processes but carries probabilistic nature so can not strictly predict whether entropy will decrease or increase in time. The problem becomes solvable in